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Analysis of Nucleating Agent (NA-27) in Polypropylene Samples By ICP MS

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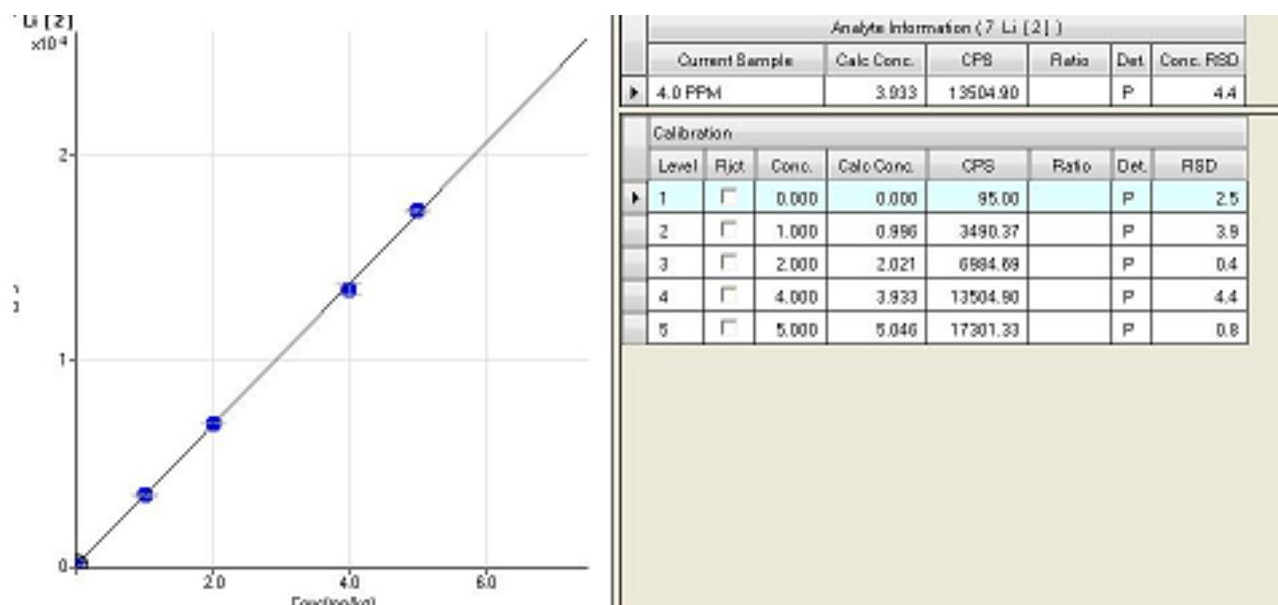
Abstract

Analysis of NA27 additive in Polypropylene samples having I-168 additive NA-27 is an advanced nucleating agent for polypropylene, providing superior mechanical properties, easier dispersion and reduced interaction with metal stearate. It is effective even in talc-filled polypropylene. Nucleating agents/Clarifiers NA 27 added in polypropylene grades (1110MAS /3120MA/ 4080MA/3650MN/3400MN) to give high degree of crystallinity resulting in increased mechanical properties such as hardness, elasticity modulus etc, and improve optical properties such as transparency. In many grades where I-168 and NA-27 are added together, it is challenging to quantify the I-168 and NA-27 additives separately. As both of the additives are phosphorus based and do not separate out in HPLC or WDXRF. To overcome this issue a new method has been developed for the analysis of NA27 contents in polypropylene samples having I-168 additives.

Keywords- HPLC, WXRF, NA-27 additives etc.

Experimental work:

ICP-MS 7700 series is used to determine the Li content in NA-27 additive. A calibration curve was prepared for Li content ranging 0.0 to 5 mg/Kg concentration.



Calibration curve of Li content

After calibration about 100 gm Polypropylene resin sample was weighted nearest 0.01 grams in a crucible and kept in muffle furnace at 600 °C for 1 hour and residue is digested with concentrated nitric acid and diluted with DM water as per desired analysis range. Digested sample was analyzed for lithium content over ICMPS operating at following conditions:

RF power	1550W
Sample depth	8 mm
Carrier gas flow	1.12 Lit/min
Nebulizer pump	0.1 rps
Rinse speed	0.50 rps
Rinse time	30 sec
Uptake time	45 sec
Uptake speed	0.50 rpm

Different standard of NA 27 were prepared (By mixing powder, NA27, I-168 weight by weight followed by single extrusion over cast line) and analyzed for Li content and respectively NA 27 concentration.

S No	Sample details	Li Concentration (mg/Kg)	Concentration NA-27 (mg/Kg)	Deviation From standard Value	% Deviation from standard
1	Pure NA 27	1.100 %			
2	PP with 0 ppm NA-27	0	0		
3	PP with 100 ppm NA-27	1.102	100	0	0.00
4	PP with 200 ppm NA-27	2.115	192	8	3.86
5	PP with 250 ppm NA-27	2.65	241	9	3.64
6	PP with 300 ppm NA-27	3.49	317	-17	-5.76
7	PP with 0350ppm NA-27	3.99	363	-13	-3.64

Repeatability of NA-27analysis in PP sample found as below:

S. No	Sample ID	Li (mg/Kg)	NA 27 (mg/Kg)
1	PP -II 879	1.75	159
2	PP -II 879	1.94	176
3	PP -II 879	2.11	192
4	PP -II 879	1.88	171
5	PP -II 879	1.96	178

Uncertainty of NA-27 measurement (at the level of 175 mg/Kg) is ± 11 mg/kg (@95% confidence level).

Conclusion-

Newly developed method is accurate and reliable for analysis of NA-27 additives. NA 27 contains two Phosphate ester compound and Organic compound. Out of two phosphate ester components of NA27, one component is around 30-40% and named as 12H- Dibenzo [d,g][1,2,3] diox aphosphocin, 2,4,8,10-tetrakis(1,1dimethylethyl)-6-hydroxy-, 6-oxide, lithium salt having a molecular formula $C_{29}H_{42}LiO_4P$ and molecular weight 492.556142 [g/mol]. Nucleating agent is added along with other additives like anti oxidants I-1010, I-168 and emulsifier GMS-90. Blends of phenolic and phosphorus based antioxidants (Like I-1010 and I-168) are analyzed by HPLC and WD XRF techniques.

References-

1. Di Bella M, Quartieri S, Sabatino G, Santalucia F, Triscari M. The glass mosaics tesserae of Villa del Casale, Piazza Armerina, Italy: A multi-technique archaeometric study. *Archaeolog Anthropolog Sci* 2014, 6, 4, 345–62. [Search in Google Scholar](#)
2. Scholze H. Glas – Natur, Struktur und Eigenschaften, Springer Verlag, Berlin Heidelberg, 1977. [Search in Google Scholar](#)
3. Varshneya AK. Fundamentals of Inorganic Glass. New York, Academic Press, Inc, Harcourt Brace & Co., 1994. [Search in Google Scholar](#)
4. Lange J. Rohstoffe der Glasindustrie, Weinheim, Wiley-VCH, 1993. [Search in Google Scholar](#)
5. Vogel W. Glaschemie. Springer Verlag, 1992. [Search in Google Scholar](#)
6. Bange K, Durán A, Parker JM. Making Glass Better – ICG Roadmaps of Glass R&D with a 25 Year Horizon, 2nd edn. Madrid, International Commission of Glass, 2014. [Search in Google Scholar](#)
7. Robertshaw, P, Wood, M, Haour, A, Karklins, K, Neff, H. Chemical analysis, chronology, and context of a European glass bead assemblage from Garumele, Niger. *J Archaeolog Sci* 2014, 41, 591–604. [Search in Google Scholar](#)

8. Siqin B, Li Q, Gan F. Analysis of ancient Chinese potash glass by laser ablation inductively coupled plasma-atomic emission spectrometry/mass spectrometry. Fenxi Huaxue 2013, 41, 9, 1328–33. [Search in Google Scholar](#)

9. Smit Z, Milavec T, Fajfar H, Rehren Th, Lankton JW, Gratuze B. Analysis of glass from the post-Roman settlement Tonovcov grad, Slovenia, by PIXE-PIGE and LA-ICP-MS. Nucl Instrum Meth Phys Res, Sect B: Beam Interact Mater At 2013, 311, 53–59. [Search in Google Scholar](#)

10. Wedepohl KH, Simon K, Kronz A. Data on 61 chemical elements for the characterization of three major glass compositions in Late Antiquity and the Middle Ages. Archaeometry 2011, 53, 1, 81–102. [Search in Google Scholar](#)

11. De Francesco AM, Scarpelli R, Barca D, Ciarallo A, Buffone L. Preliminary chemical characterization of Roman glass from Pompeii. Periodico di Mineralogia 2010, 79, 3, 11–19. [Search in Google Scholar](#)

12. Robertshaw P, Benco N, Wood M, Dussubieux L, Melchiorre E, Ettahiri A. Chemical analysis of glass beads from medieval Al-Basra (Morocco). Archaeometry 2010, 52, 3, 355–79. [Search in Google Scholar](#)

13. Smit Z, Janssens K, Bulska E, Wagner B, Kos M, Lazar I. Trace element fingerprinting of Facon-de-Venise glass. Nucl Instrum Meth Phys Res, Sect B: Beam Interact Mater At 2005, 239, 1–2, 94–99. [Search in Google Scholar](#)

14. De Raedt I, Janssens K, Veeckman J, Vincze L, Vekemans B, Jeffries TE. Trace analysis for distinguishing between Venetian and facon-de-Venise glass vessels of the 16th and 17th century. J Anal Atom Spectrom 2001, 16, 9, 1012–17. [Search in Google Scholar](#)

15. Scarpelli R, DeFrancesco AM, Gaeta M, Cottica D, Toniolo L. The provenance of the Pompeii cooking wares: Insights from LA-ICP-MS trace element analyses. Microchem J 2015, 119, 93–101. [Search in Google Scholar](#)

16. Wang X, Motto-Ros V, Panczer G, et al. Mapping of rare earth elements in nuclear waste glass-ceramic using micro laser-induced breakdown spectroscopy. Spectrochim Acta, Part B: At Spectrosc 2013, 87, 139–46. [Search in Google Scholar](#)
17. Tabersky D, Luechinger NA, Rossier M, et al. Development and characterization of custom-engineered and compacted nanoparticles as calibration materials for quantification using LA-ICP-MS. J Anal At Spectrom 2014, 29, 6, 955–62. [Search in Google Scholar](#)
18. Tu XL, Zhang H, Deng WF, et al. Application of Resolution in-situ laser ablation ICP-MS in trace element analyses. Diqui Huaxue 2011, 40, 1, 83–98. [Search in Google Scholar](#)
19. Strnad L, Mihaljevic M, Sebek O. Laser ablation and solution ICP-MS determination of rare earth elements in USGS BIR-1G, BHVO-2G and BCR-2G glass reference materials. Geostand Geoanal Res 2005, 29, 3, 303–14. [Search in Google Scholar](#)
20. Hu Z, Gao S, Liu Y, Chen H, Hu S. Accurate determination of rare earth elements in USGS, NIST SRM, and MPI-DING glasses by excimer LA-ICP-MS at high spatial resolution. Spectrosc Lett 2008, 41, 5, 228–36. [Search in Google Scholar](#)
21. Jochum KP, Stoll B, Herwig K, et al. MPI-DING reference glasses for in situ microanalysis: new reference values for element concentrations and isotope ratios. Geochem, Geophys, Geosyst 2006, 7, 2. [Search in Google Scholar](#)
22. Eggins SM, Shelley JM. Compositional heterogeneity in NIST SRM 610-617 glasses. Geostand Newsl 2002, 26, 3, 269–86. [Search in Google Scholar](#)
23. Horn I, Hinton RW, Jackson SE, Longerich HP. Ultra-trace element analysis of NIST SRM 616 and 614 using laser ablation microprobe-inductively coupled plasma-mass spectrometry (LAM-ICP-MS): a comparison with secondary ion mass spectrometry (SIMS). Geostand Newsl 1997, 21, 2, 191–203. [Search in Google Scholar](#)

24. Norman MD, Pearson NJ, Sharma A, Griffin WL. Quantitative analysis of trace elements in geological materials by laser ablation ICPMS: Instrumental operating conditions and calibration values of NIST glasses. *Geostand News* 1996, 20, 2, 247–61. [Search in Google Scholar](#)
25. Payne JL, Pearson NJ, Grant KJ, Halverson GP. Reassessment of relative oxide formation rates and molecular interferences on in situ lutetium-hafnium analysis with laser ablation MC-ICP-MS. *J Anal At Spectrom* 2013, 28, 7, 1068–79. [Search in Google Scholar](#)
26. Stix J, Gauthier G, Ludden JN. A critical look at quantitative laser-ablation ICP-MS analysis of natural and synthetic glasses. *Can Mineral* 1995, 33, 2, 435–44. [Search in Google Scholar](#)
27. Zhang J, Xie F, Liang Y. The impact of the mechanical properties with rare earth elements on the black shale glass-ceramic. *International Conference on Materials for Renewable Energy and Environment*, Chengdu, China 2013, 2, 569–72. [Search in Google Scholar](#)
28. Mirti P, Pace M, Ponzi MM, Aceto M. ICP-MS analysis of glass fragments of Parthian and Sasanian epoch from Seleucia and Veh Ardasir (central Iraq). *Archaeometry* 2008, 50, 3, 429–50. [Search in Google Scholar](#)
29. Smit Z, Janssens K, Bulska E, Wagner B, Kos M, Lazar I. Trace element fingerprinting of Facon-de-Venise glass. *Nucl Instrum Methods Phys Res Sect B: Beam Interact Mater At* 2005, 239, 1–2, 94–99. [Search in Google Scholar](#)
30. Mizusuna Hirobumi, Nakayama Mokichi. Determination of rare earth elements in glass by ICP-AES (inductively coupled plasma-Atomic emission spectroscopy). *Jpn Kokai Tokkyo Koho* 1993, JP05079983A19930330. [Search in Google Scholar](#)
31. Abe Y, Kikugawa T, Nakai I. Development of nondestructive heavy elemental analytical method of ancient glass artefacts using high-energy (116 keV) synchrotron radiation X-ray fluorescence spectrometry. *X-sen Bunseki no Shinpo* 2014, 45, 251–68. [Search in Google Scholar](#)

32. Chen JR, Chao EC, Back JM, et al. Rare earth element concentrations in geological and synthetic samples using synchrotron x-ray fluorescence analysis. Nucl Instrum Methods Phys Res Sect B: Beam Interact Mater At 1993,B75,1–4, 576–81. [Search in Google Scholar](#)
33. Nikkei I, Terada Y, Itou M, Sakurai Y. Use of highly energetic (116 keV) synchrotron radiation for X-ray fluorescence analysis of trace rare-earth and heavy elements. J Synchrotron Radiat 2001, 8, 4, 1078–81. [Search in Google Scholar](#)
34. Nakanishi T, Nishiwaki Y, Miyamoto N, Shimoda O, Watanabe S, Muratsu S, Takatsu M, Terada Y, Suzuki Y, Kasamatsu M, Suzuki S. Lower limits of detection of synchrotron radiation high-energy X-ray fluorescence spectrometry and its possibility for the forensic application for discrimination of glass fragments. Forensic Sci Int 2008, 175, 2–3, 227–34. [Search in Google Scholar](#)
35. Nishiwaki Y, Nakanishi T, Terada Y, Ninomiya T, Nakai I. Nondestructive discrimination of small glass fragments for forensic examination using high energy synchrotron radiation X-ray fluorescence spectrometry. X-Ray Spectrom 2006, 35, 3, 195–99. [Search in Google Scholar](#)
36. Baklouti S, Maritan L, Laridhi Ouazaa N, Mazzoli C, Larabi Kassaa S, Joron JL Fouzai B, Casas Duocastella L, Labayed-Lahdari M. African terra sigillata from Henchir Es-Srira archaeological site, central Tunisia: Archaeological provenance and raw materials based on chemical analysis. Appl Clay Sci 2015, 105–106, 27–40. [Search in Google Scholar](#)
37. Facetti-Masulli JF, Kump P, Gonzalez VR, Diaz Z. Geochemical studies of Guarani ethnic groups pottery with XRF. J Radioanal Nucl Chem 2010, 286, 2, 489–94. [Search in Google Scholar](#)
38. Beckhoff B, Kolbe M, Hahn O, Karydas AG, Zarkadas C, Sokaras D, Mantler M. Reference-free x-ray fluorescence analysis of an ancient Chinese ceramic. X-Ray Spectrom 2008, 37, 4, 462–65. [Search in Google Scholar](#)
39. Bach H, Krause D. Analysis of the Composition and Structure of Glass and Glass Ceramics. Berlin, Heidelberg, Springer Verlag, 2013, 83. [Search in Google Scholar](#)

40. Deutsche Norm DIN 52340–2. Prüfung von Glas; Chemische Analyse von ungefärbten Kalk-Natron-Gläsern, Teil 2: Bestimmung von SiO₂, 1974. [Search in Google Scholar](#)
41. International Standard ISO 21078–1. Determination of boron (III) oxide in refractory products – Part 1: Determination of total boron (III) oxide in oxidic materials for ceramics, glass and glazes, 2008. [Search in Google Scholar](#)
42. American Standard ASTM C169–92. Standard test methods for chemical analysis of soda-lime and borosilicate glass, 2011. [Search in Google Scholar](#)
43. Deutsche Norm DIN 51084. Prüfung von oxidischen Roh- und Werkstoffen für Keramik, Glas und Glasuren – Bestimmung des Gehaltes an Fluorid (Testing of oxidic raw and basic materials for ceramic, glass and glazes – Determination of fluoride content) – in German, 2008. [Search in Google Scholar](#)
44. Deutsche Norm DIN 52340–3. Prüfung von Glas; Chemische Analyse von ungefärbten Kalk-Natron-Gläsern, Teil 3: Aufschlußverfahren, 1990. [Search in Google Scholar](#)
45. Deutsche Norm DIN 51086–2. Prüfung von oxidischen Roh- und Werkstoffen für Keramik, Glas und Glasuren – Teil 2: Bestimmung von Ag, As, B, Ba, Be, Bi, Ca, Cd, Ce, Co, Cr, Cu, Er, Eu, Fe, La, Mg, Mn, Mo, Nd, Ni, P, Pb, Pr, S, Sb, Se, Sn, Sr, Ti, V, W, Y, Yb, Zn, Zr durch optische Emissionsspektrometrie mit induktiv gekoppeltem Plasma (ICP OES) (Testing of oxidic raw materials and materials for ceramics, glass and glazes – Part 2: Determination of Ag, As, B, Ba, Be, Bi, Ca, Cd, Ce, Co, Cr, Cu, Er, Eu, Fe, La, Mg, Mn, Mo, Nd, Ni, P, Pb, Pr, S, Sb, Se, Sn, Sr, Ti, V, W, Y, Yb, Zn, Zr by optical emission spectrometry with inductively coupled plasma (ICP OES)) – in German, 2004. [Search in Google Scholar](#)
46. Borom MP, Hanneman RE. Local compositional changes in alkali silicate glasses during electron microprobe analysis. J Appl Phys 1967, 38, 2406. [Search in Google Scholar](#)
47. Spray JG, Rae DA. Quantitative electron microprobe analysis of alkali silicate glasses: A review and userguide. Can Mineral 1995, 33, 323–32. [Search in Google Scholar](#)

48. Varshneya AK, Cooper AR, Cable M. Changes in composition during electron micro-probe analysis of K₂O–SrO–SiO₂ Glass. J Appl Phys 1966, 37, 2199. [Search in Google Scholar](#)
49. Nielsen CH, Sigurdsson H. Quantitative methods for electron microprobe analysis of sodium in natural and synthetic glasses. Am Mineral 1981, 66, 547–52. [Search in Google Scholar](#)
50. Vassamillet LF, Caldwell VE. Electron probe microanalysis of alkali metals in glasses. J Appl Phys 1969, 40, 1637. [Search in Google Scholar](#)
51. Williams DB, Carter CB. Transmission Electron Microscopy: Basics I. New York, Springer Science & Business Media Inc., 1996. [Search in Google Scholar](#)
52. Agrawal A & Kumar P , Effect of Storage Time Over FM and Izod Property of PP Resins, International Journal of Multidisciplinary Research Education Analysis and Development (IJMREAD) Delhi, 2021,2,06-19.

